

0040-4020(94)00838-8

Geminal Dialkylation, Alkylative Reduction and Olefination of Aliphatic Aldehydes. Reaction of gem-Bistriflates with Higher Order Dialkylcyanocuprates.

A. García Martínez,^{a*} J. Osío Barcina,^a B. Ruiz Díez,^a L. R. Subramanian^b

^aDepartamento de Química Orgánica, Facultad de Ciencias Químicas, Úniversidad Complutense, Ciudad Universitaria, E-28040 Madrid, Spain

^bInstitut für Organische Chemie, Lehrstuhl für Organische Chemie II, Auf der Morgenstelle 18, D-72076 Tübingen, Germany

Abstract: gem-Dialkylation or alkylative reduction of α -unbranched aliphatic aldehydes 1 is advantageously achieved by reaction of the corresponding gem-bistriflates 2 with di-n-alkylcyanocuprates or di-sec- and di-tert-alkylcyanocuprates respectively. The reaction of α -branched gem-bistriflates 2 with dialkylcyanocuprates in the presence of boron trifluoride affords the olefins 6 in good yield.

The gem-bistriflates 2 are easily obtained in good yields as primary products in the reaction of aliphatic aldehydes 1 with triflic anhydride (Tf_2O) .^{1,2} The chemistry of these compounds is important because they can be considered as synthetic equivalents of the dications 3 (Scheme 1). In search of new methods for the formation of carbon framework, we present here the reaction of gem-bistriflates 2 with organocuprates. Lithium dialkylcuprates couple with vinyl triflates, however, they do not react with aryl triflates.³ Coupling reactions of higher order dialkylcyanocuprates⁴ with both vinyl⁵ and aryl triflates,⁶ but not with alkyl triflates, have been reported.

We have found that the reaction of lithium dialkylcuprates with 2 affords always complicated mixtures of products. Fortunately, the reaction with higher order dialkylcyanocuprates $R_2^{3}(CN)CuLi_2$ in tetrahydrofuran furnishes single products in good yields, whose structure is very much dependent on the steric hindrance of the radicals R^{1} - R^{3} (Scheme 1 and Table).

Thus, the reaction of α -unbranched *gem*-bistriflates 2 (R² = H) with primary cyanocuprates (R³ = CH₃, *n*-Bu) yields the corresponding *gem*-dialkylated products 5a-d. When R³ is secondary (R³ = s-Bu) or tertiary (R³ = t-Bu) the reaction leads to the alkylative reduction products 4e, f. In the case of α -branched 2 (R¹, R² \neq H), the reaction takes place only when R³ = CH₃ yielding the olefins 6g, k, m, q. However, under addition of Et₂O.BF₃⁶ the reaction succeeds also with secondary and tertiary radicals (R³ = s-Bu, t-Bu) affording



Scheme 1

2	Product	Yield (%) ^a
	5a	60
	5b	50
	5c	70
	5d	40
	4 e	56
	4f	44
πọ		
	бд	$80 (Z/E = 55/45)^{b}$
	6h	$52 (Z/E = 55/45)^{b}$
	6i	$71 (Z/E = 55/45)^{b}$
	бј	$60 (Z/E = 52/48)^{b}$
	бк	50 (Z/E= 74/26) ^b
	61	55 ($Z/E = 60/40$) ^b
\sim	6ா	85
011	6n	50
-	60	77
	бр	40
	6 q	60

Table. Reaction of gem-Bistriflates 2 with Higher Order Dialkylcyanocuprates.

^aYield of isolated product. ^bCalculated by GC. The configurations were determinated by ¹³C-NMR.

the corresponding olefins **6h-j**, **l**, **n-p**. It is noteworthy that, although in some examples (**6g-6j**) the Z/E ratios are close to 50:50, the Z-isomers were obtained predominantly in every case (see Table) and also that this procedure leads to olefins which are positional isomers of those obtained with the Wittig and related reactions⁷ from the same aldehydes (Scheme 2).



Scheme 2

In summary, the reaction of α -unbranched gem-bistriflates 2 with higher order dialkylcyanocuprates is an efficient alternative to the gem-dialkylation of α -unbranched aldehydes via the corresponding gemdihalides.⁸ The reaction of α -branched gem-bistriflates 2 constitutes a new method for the regioselective olefination of the corresponding aldehydes, which is suitable for the preparation of highly hindered alkenes. Studies on the mechanism and the influence of promoters to increase the stereoselectivity of the process are in progress.

EXPERIMENTAL

Starting materials and solvents were used as received from commercial suppliers. Tetrahydrofuran (THF) was distilled from sodium/benzophenone and LiAlH₄ under argon atmosphere. ¹H-NMR and ¹³C-NMR spectra were recorded on a Bruker-AC 250 MHz spectrometer in CDCl₃ and chemical shifts are expressed in ppm. IR spectra were recorded on a Perkin-Elmer 781 spectrometer. Mass spectra were recorded on a Varian-MAT 711 instrument.

General procedure for the preparation of gem-bistriflates. The gem-bistriflates 2 were prepared according to a procedure previously described by us,¹ by reaction of the corresponding aldehyde (25 mmol) with trifluoromethanosulphonic anhydride (Tf₂O) (27 mmol) in CH₂Cl₂ (100 ml) at 0°C. Yield 60-85%.

General procedure for the reaction of gem-bistriflates 2 with higher order lithium dialkylcyanocuprates. To a suspension of CuCN (4 mmol) in 50 ml of THF at -78°C under argon was added the corresponding alkyllithium (8 mmol) dropwise with stirring. The mixture was allowed to warm to 0°C for 10 min and cooled to -78°C. A solution of the gem-bistriflate 2 (1 mmol) in THF (10 ml) was added slowly. After stirring for

12 h the reaction mixture was hydrolyzed with a 10% solution of NH_4OH in saturated NH_4Cl (100 ml). The resulting mixture was filtered through Celite, extracted with pentane (3 x 25 ml) and dried over magnesium sulphate. The solvent was removed by distillation and the reaction products were purified by column chromatography (silica gel, *n*-pentane). When the reaction was carried out with α -branched gem-bistriflates 2, 4 mmol of boron trifluoride etherate were added to the solution of the dialkylcyanocuprate just before the addition of the triflate.

Data of 5a:⁸ Yield: 60%. IR (film): $\nu = 2970$, 2930, 1470, 1380, 1370 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 1.56$ -1.42 (m, 1H), 1.24 (m, 12H), 0.86 (t, 3H), 0.84 (d, 6H) ppm; ¹³C-NMR (CDCl₃): $\delta = 39.0$, 31.9, 29.8, 29.3, 27.9, 27.3, 22.6, 22.6, 14.0 ppm.

Data of 5b: Yield: 50%. IR (CCl₄): $\nu = 2970$, 2930, 2860, 1470, 1390, 1370 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 1.42$ -1.56 (m, 1H), 1.24 (m, 22H), 0.86 (t, 3H), 0.84 (d, 6H) ppm; ¹³C-NMR (CDCl₃): $\delta = 39.1$, 32.0, 30.0, 29.7, 29.4, 28.0, 27.5, 22.8, 22.7, 14.2 ppm; MS m/e (%B): 212 (M⁺, 9), 169 (12), 113 (14), 99 (19), 71 (52), 57 (100), 43 (98).

Data of 5c:⁸ Yield: 70%. IR (CCl₄): ν = 3020, 2970, 2870, 1600, 1500, 1450, 1390, 1370, 750, 700 cm⁻¹; ¹H-NMR (CDCl₃): δ = 7.43 (m, 5H), 2.75 (t, 2H), 1.8-1.6 (m, 3H), 1.08 (d, 6H) ppm; ¹³C-NMR (CDCl₃): δ = 143.2, 128.4, 128.3, 125.6, 41.66, 34.6, 22.6, 28.4 ppm.

Data of 5d:⁸ Yield: 40%. IR (CCl₄): ν = 3020, 2970, 2930, 2870, 1600, 1490, 1460, 1380 cm⁻¹; ¹H-NMR (CDCl₃): δ = 7.3-7.1 (m, 5H), 2.6-2.5 (t, 2H), 1.6-1.5 (m, 1H), 1.3-1.2 (m, 14H), 0.89 (m, 6H) ppm; ¹³C-NMR (CDCl₃): δ = 143.4, 128.4, 128.3, 125.6, 37.2, 35.8, 33.2, 28.9, 23.2, 14.3 ppm.

Data of 4e:⁸ Yield: 56%. IR (CCl₄): ν = 3020, 2970, 2930, 2870, 1600, 1500, 1450, 1375, 730, 700 cm⁻¹; ¹H-NMR (CDCl₃): δ = 7.28-7.06 (m, 5H), 2.6 (t, 2H), 1.7-1.6 (m, 2H), 1.3 (m, 3H), 1.2-1.1 (m, 2H), 0.8 (m, 6H) ppm; ¹³C-NMR (CDCl₃): δ = 142.9, 128.3, 128.2, 125.5, 36.4, 34.2, 29.2, 28.8, 19.0, 11.5 ppm.

Data of 4f:⁸ Yield: 44%. IR (CCl₄): ν = 3020, 2970, 2870, 1600, 1500, 1450, 1390, 1360, 1250, 900 cm⁻¹; ¹H-NMR (CDCl₃): δ = 7.3-7.1 (m, 5H), 2.55 (t, 2H), 1.7-1.5 (m, 2H), 1.2 (m, 2H), 0.85 (s, 9H) ppm; ¹³C-NMR (CDCl₃): δ = 143.1, 128.5, 128.3, 125.7, 44.1, 37.0, 30.4, 29.7, 26.8 ppm.

Data of Z/E-6g:⁹ Yield: 80%. IR (film): $\nu = 2970, 2930, 1670, 1460, 1380, 910, 830, 750 cm⁻¹; ¹H-NMR (CDCl₃): <math>\delta = 5.2$ -5.1 (q, 1H), 2.15-1.8 (m, 4H), 1.6-1.5 (d, 3H), 1.3-1.2 (m, 5H), 0.9-0.8 (m, 5H) ppm; ¹³C-NMR (CDCl₃): $\delta = 142.1, 117.5$ (E), 116.9 (Z), 36.5, 30.6, 30.5, 29.7, 29.6, 22.9, 22.7, 22.6, 14.1,

14.0, 13.2, 13.0, 12.9, 12.8 ppm; MS m/e (%B): 126 (M⁺, 9), 84 (25), 69 (51), 56 (20), 55 (100), 43 (13), 41 (58).

Data of Z/E-6h:¹⁰ Yield: 52%. IR (film): $\nu = 2990$, 2930, 2870, 1660, 1455, 1380, 1010, 850, 730 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 5.1$ (q, 1H), 2.0 (m, 6H), 1.3-1.2 (m, 7H), 0.9-0.7 (m, 10H) ppm; ¹³C-NMR (CDCl₃): $\delta = 141.1$, 141.1, 124.1 (E), 123.4 (Z), 36.3, 32.5, 32.0, 30.8, 30.5, 29.9, 29.6, 29.4, 27.4, 27.3, 23.0, 22.9, 22.7, 22.6, 22.5, 22.4, 14.1, 14.0, 13.2, 12.9 ppm; MS m/e (%B): 168 (M⁺, 12), 111 (6), 97 (19), 83 (37), 69 (100), 55 (85), 41 (62).

Data of Z/E-6i: Yield: 71%. IR (film): $\nu = 2980$, 2970, 2890, 1660, 1460, 1380, 890 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 6.4$ (d, 1H), 2.3-2.2 (m, 1H), 2.1 (m, 4H), 1.4-1.2 (m, 10H), 0.8-1 (m, 8H) ppm; ¹³C-NMR (CDCl₃): $\delta = 135.1$, 130.6 (E), 130.3 (Z), 34.2, 30.2, 29.8, 29.3, 27.2, 24.3, 22.5, 22.4, 22.3, 20.6, 14.1, 13.8, 12.1, 11.9 ppm; MS m/e (%B): 168 (M⁺, 11), 139 (19), 111 (17), 97 (17), 83 (82), 69 (71), 55 (100), 41 (54).

Data of Z/E-6j:¹⁰ Yield: 60%. IR (film): $\nu = 2970$, 2940, 2870, 1650, 1460, 1360, 910, 840 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 5.1$ (s, 1H), 2.1 (q, 2H), 1.9 (q, 2H), 1.5-1.4 (m, 5H), 1.1 (s, 9H), 0.8-0.7 (m, 5H) ppm; ¹³C-NMR (CDCl₃): $\delta = 140.7$, 140.4, 134.5 (E), 133.6 (Z), 37.2, 34.2, 31.7, 31.6, 31.1, 31.0, 30.7, 30.3, 23.3, 23.1, 22.5, 22.4, 14.2, 14.1, 13.5, 13.4 ppm; MS m/e (%B): 168 (M⁺, 8), 139 (7), 111 (23), 97 (25), 83 (51), 69 (100), 55 (66), 41 (56).

Data of Z/E-**6k**:¹¹ Yield: 50%. IR (film): $\nu = 3020, 2970, 2920, 2870, 1600, 1500, 1450, 1380, 1040, 910, 820, 760, 700 cm⁻¹; ¹H-NMR (CDCl₃): <math>\delta = 7.4$ -7.2 (m, 5H), 5.6 (q, 1H), 2.1 (dq, 3H), 1.6 (dq, 3H) ppm; ¹³C-NMR (CDCl₃): $\delta = 141.9$ (Z), 136.8 (E), 128.5, 127.7, 126.4, 121.7, 25.5 (Z), 22.3 (E), 14.9 (Z), 14.0 (E) ppm; MS m/e (%B): 132 (M⁺, 64), 117 (29), 91 (39), 77 (16), 65 (21), 51 (28).

Data of *Z*/*E***-61**: Yield: 55%. IR (film): $\nu = 2970$, 2940, 1650, 1460, 1360, 910, 740 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 5.15$ (s, 1H), 1.9-1.7 (m, 1H), 1.6-1.5 (s, 3H), 1.4-1.2 (m, 2H), 1.0 (s, 9H), 0.8-0.6 (m, 6H) ppm; ¹³C-NMR (CDCl₃): $\delta = 139.3$, 135.7 (*E*), 134.7 (*Z*), 46.2, 35.8, 32.3, 31.9, 31.2, 27.7, 27.6, 22.5, 19.7, 19.1, 18.7, 12.8, 12.3, 12.1 ppm; MS m/e (%B): 154 (M⁺, 10), 125 (6), 97 (56), 83 (62), 69 (100), 55 (82), 41 (64).

Data of 6m:¹² Yield: 85%. IR (film): $\nu = 2940$, 2870, 1645, 1450, 1387, 1010, 900, 820 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 5.20$ (qd, 1H), 1.9 (m, 4H), 1.40 (d, 3H), 1.39-1.35 (m, 6H) ppm; ¹³C-NMR (CDCl₃): $\delta =$

140.2, 115.1, 37.2, 28.5, 28.1, 27.5, 26.9, 12.6 ppm; MS m/e (%B): 110 (M⁺, 23), 95 (14), 82 (22), 81 (100), 69 (22), 67 (64), 41 (45).

Data of 6n:¹³ Yield: 50%. IR (film): $\nu = 2970$, 2920, 1650, 1470, 1385, 1345, 1200, 1090, 1050 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 5.1$ -5.0 (t, 1H), 2.1-1.9 (m, 6H), 1.6-1.5 (m, 6H), 1.2 (m, 4H), 0.9-0.8 (t, 3H) ppm; ¹³C-NMR (CDCl₃): $\delta = 139.5$, 121.4, 37.2, 32.4, 28.7, 28.6, 27.8, 26.9, 26.7, 22.9, 14.0 ppm; MS m/e (%B): 152 (M⁺, 13), 109 (29), 96 (29), 81 (53), 67 (100), 55 (26), 41 (34).

Data of 60: Yield: 77%. IR (film): $\nu = 2960$, 2860, 1665, 1450, 1370, 790 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 4.8$ (d, 1H), 2.3 (m, 1H), 2.2-2.0 (m, 4H), 1.5 (m, 6H), 1.3-1.1 (m, 2H), 1.0-0.8 (m, 6H) ppm; ¹³C-NMR (CDCl₃): $\delta = 138.3$, 128.1, 37.4, 33.5, 30.7, 29.2, 28.9, 28.1, 26.9, 21.5, 12.1 ppm; MS m/e (%B): 152 (M⁺, 17), 123 (81), 95 (12), 81 (100), 67 (80), 41 (44), 28 (78), 17 (82).

Data of 6p:^{13,14} Yield: 40%. IR (film): $\nu = 2970$, 2930, 1655, 1450, 1370, 1205, 840 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 5.1$ (s, 1H), 2.2. (s, 2H), 1.95 (s, 2H), 1.5 (m, 6H), 1.1 (s, 9H) ppm; ¹³C-NMR (CDCl₃): $\delta = 139.4$, 131.9, 38.7, 31.6, 29.9, 29.8, 29.1, 27.8, 27.3 ppm; MS m/e (%B): 152 (M⁺, 16), 137 (25), 109 (30), 95 (19), 81 (43), 69 (100), 55 (29), 41 (47).

Data of 6q:¹² Yield: 60%. IR (film): $\nu = 2930$, 2870, 1640, 1470, 1450 cm⁻¹; ¹H-NMR (CDCl₃): $\delta = 5.3-5.2$ (qd, 1H), 2.2-2.1 (m, 7H), 1.6-1.5 (m, 10H) ppm; ¹³C-NMR (CDCl₃): $\delta = 141.5$, 118.7, 37.7, 28.5, 27.3, 27.1, 26.6, 26.5, 26.0, 25.9 ppm; MS m/e (%B): 138 (M⁺, 92), 123 (11), 110 (96), 109 (96), 95 (67), 82 (98), 67 (100).

Acknowledgment: We thank the DGICYT (Spain) for finantial support of this work (Grant PB90-0070).

REFERENCES

- 1. García Martínez, A.; Martínez Alvarez, R.; García Fraile, A.; Subramanian, L. R.; Hanack, M. Synthesis 1987, 49.
- 2. Wright, M. E.; Pulley, S. R. J. Org. Chem. 1989, 54, 2886.
- 3. Lipshutz, B. E. Synlett 1990, 119.
- 4. McMurry, J. E.; Mohanraj, S. Tetrahedron Lett. 1983, 24, 2723.

- 5. Lipshutz B. E.; Elworthy, T. R. J. Org Chem. 1990, 55, 1695.
- The use of Et₂O.BF₃ as promoter in coupling reactions is described in: Lipshutz, B. H.; Ellsworth, E. L.; Siahaan, T. J. J. Am. Chem. Soc. 1988, 110, 4834.
- 7. March, J. Advanced Organic Chemistry; Wiley: New York, 1992.
- García Martínez, A.; Herrera Fernández, A.; Martínez Alvarez, R.; Osío Barcina, J.; García Gómez, C.; Subramanian, L. R. Synthesis 1993, 1063.
- 9. Barluenga, J.; Yus, M.; Concellón, J. M. J. Org. Chem. 1983, 48, 609.
- 10. Commercon, A.; Normant, J.F.; Villieras, J. J. Organomet. Chem. 1977, 128, 1.
- 11. Sharpless, K.B.; Young, M.W.; Lauer, R. Tetrahedron Lett. 1973, 22, 1979.
- 12. Mach, K.; Turecex, F.; Antropiusova, H.; Petrusova, L.; Hanus, V. Synthesis 1982, 1, 53.
- 13. Zweifel, G.; Fisher, R.P. Synthesis 1972, 10, 557.
- 14. Herberg, C.; Beckhaus, H.D.; Körtvelyesi, T.; Rücherdt, C. Chem. Ber. 1993, 126, 117.

(Received in UK 25 July 1994; revised 26 September 1994; accepted 30 September 1994)